

Stereoselective Syntheses of (\pm) -Komaroviquinone and (±)-Faveline Methyl Ether through Intramolecular Heck Reaction

Sujaya Sengupta,† Michael G. B. Drew,‡ Ranjan Mukhopadhyay,† Basudeb Achari,† and Asish Kr. Banerjee*,†

Medicinal Chemistry Division, Indian Institute of Chemical Biology, 4 Raja S. C. Mullick Road, Kolkata-700 032, India, and School of Chemistry, University of Reading, Whiteknights, Reading RG6 6AD, U.K.

ashisbanerjee@iicb.res.in

Received May 30, 2005

An efficient, flexible, and stereoselective convergent route for constructing the trans-10-hydroxy-1,1-dimethyloctahydrodibenzo[a,d]cyclohepten-7-ones ($\mathbf{5a-c}$) was achieved via intramolecular Heck reaction. This strategy has been successfully implemented for the syntheses of (\pm) -komaroviquinone (3) through (\pm) -coulterone dimethyl ether (5c) and (\pm) -faveline methyl ether (1a).

Introduction

A number of cytocidal compounds have been isolated from the bark of the Brazilian plant favela (Cnidoscolus phyllacanthus).1 Faveline methyl ether (1a), faveline (1b), and deoxofaveline (1c) are typical members of rearranged 9(10→20)-abeo-16,17-dinorabieta-8,11,13triene diterpenes (Figure 1). These molecules display significant activity against P-388 murine leukemia cells.¹ Several other rearranged abietane diterpenes, for example, pisiferin (1d), ^{2a,b} isopisiferin (1e), ^{2c} and barbatusol (1f),^{2d} have also been described. Recently, isolation of coulterone $(2)^{3,4}$ and two new compounds in the icetexane group of rearranged abietane diterpenes komaroviquinone $(3)^4$ and cyclocoulterone $(4)^4$ have been reported from Dracocephalum komarovi. Komaroviquinone (3) displays a strong trypanocidal activity against epimastigotes of Trypanosoma cruzi, the causative agent of American trypanosomiasis.4

A number of total syntheses of the rearranged diterpenoids 1d,⁵ 1e,^{5b,6} and 1f^{5c,7} incorporating functionalized 6-7-6-fused tricycles have appeared in the literature. The reported total syntheses^{5c,8} of (\pm) -faveline methyl ether (1a), (\pm) -faveline (1b), and (\pm) -deoxofaveline (1c)involve generation of the basic tricyclic skeleton mainly through acid-catalyzed cycliacylation⁸ or cyclialkylation^{5c} reactions. However, introduction of the 7-oxo group in deoxofaveline (1c) was plagued with problems. 5c,8b No report exists so far for the synthesis of coulterone (2), komaroviquinone (3), and cyclocoulterone (4). In continuation of our studies on the palladium-mediated synthesis of rearranged polycyclic diterpenoids, 9,10 we report herein the first total synthesis of (\pm) -komaroviguinone (3) through (\pm) -coulterone dimethyl ether (5c) as well as a simple route to (\pm) -faveline methyl ether (1a).

33, 655–658. (b) Deb, S.; Bhattacharjee, G.; Ghatak, U. R. J. Chem. Soc., Perkin Trans. 1 1990, 1453–1458.

[†] Indian Institute of Chemical Biology.

[‡] University of Reading. (1) Endo, Y.; Ohta, Y.; Nozoe, S. *Tetrahedron Lett.* **1991**, *32*, 3083–

^{(2) (}a) Yatagai, M.; Takahashi, T. Phytochemistry 1980, 19, 1149–1151. (b) Hasegawa, S.; Hirose, Y.; Yatagai, M.; Takahashi, T. Chem. Lett. 1984, 1837-1838. (c) Hasegawa, S.; Kojima, T.; Hirose, Y. Phytochemistry 1985, 24, 1545–1551. (d) Kelecom, A. Tetrahedron 1983, 39, 3603 - 3609.

⁽³⁾ Frontana, B.; Cárdenas, J.; Rodríguez-Hahn, L. Phytochemistry **1994**, 36, 739-741.

⁽⁴⁾ Uchiyama, N.; Kiuchi, F.; Ito, M.; Honda, G.; Takeda, Y.; Khodzhimatov, O. K.; Ashurmetov, O. A. J. Nat. Prod. 2003, 66, 128-

^{(5) (}a) Matsumoto, T.; Imai, S.; Yoshinari, T.; Matsuno, S. Bull. Chem. Soc. Jpn. 1986, 59, 3103-3108. (b) Kametani, T.; Kondoh, H.; Tsubuki, M.; Honda, T. J. Chem. Soc., Perkin Trans. 1 1990, 5–10. (c) Majetich, G.; Hicks, R.; Zhang, Y.; Tian, X.; Feltman, T. L.; Fang, J.;
Duncan, S., Jr. J. Org. Chem. 1996, 61, 8169-8185.
(6) (a) Ghosh, A. K.; Ray, C.; Ghatak, U. R. Tetrahedron Lett. 1992,

^{(7) (}a) Koft, E. R. Tetrahedron 1987, 43, 5775–5780. (b) Majetich, G.; Zhang, Y.; Feltman, T. L.; Duncan, S., Jr. Tetrahedron Lett. 1993, 34.445 - 448.

^{(8) (}a) Ghosh, A. K.; Ray, C.; Ghatak, U. R. *Tetrahedron Lett.* **1992**, 33, 655–658. (b) Ghosh, A. K.; Mukhopadhyay, C.; Ghatak, U. R. *J. Chem. Soc.*, *Perkin Trans.* 1 **1994**, 327–332.

⁽⁹⁾ Banerjee, M.; Mukhopadhyay, R.; Achari, B.; Banerjee, A. Kr. Org. Lett. **2003**, *5*, 3931–3933.

⁽¹⁰⁾ Sengupta, S.; Mukhopadhyay, R.; Achari, B.; Banerjee, A. Kr. Tetrahedron Lett. 2005, 46, 1515–1519.

FIGURE 1. Rearranged 9(10→20)-abeo-abietane diterpenoids.

SCHEME 1. Retrosynthetic Analysis

a: $R_1=R_2=R_3=H$; **b**: $R_1=R_3=H$, $R_2=Me$; **c**: $R_1=R_3=OMe$, $R_2=^{i}Pr$

Results and Discussion

A short, flexible, and stereoselective convergent route for the construction of the intermediate trans-10-hydroxy-1,1-dimethyloctahydrodibenzo[a,d]cyclohepten-7-one core structures ($\mathbf{5a}-\mathbf{c}$) was developed by employing the intramolecular Heck reaction as the crucial carbon—carbon bond-forming step, and the retrosynthetic analysis of the tricyclic systems is shown in Scheme 1.

The key step involves the formation of the cycloheptene ring $({\bf 6a-c})$ by a palladium-catalyzed 7-exo cyclization¹¹ of the allyl o-bromobenzylcyclohexanols $({\bf 7a-c})$ with the exo-methylene group in the correct position for oxidative transformation to the hydroxycycloheptenones $({\bf 5a-c})$. The desired olefinic substrates ${\bf 7a-c}$ for the intramolecular Heck reaction were prepared from the corresponding benzylcyclohexanols ${\bf 8a-c}$, which were obtained by Barbier coupling ${\bf 12a,b}$ of the benzyl chlorides ${\bf 10a-c}$

with the allylcyclohexanone **9**. The chlorides **10b** and **10c** were prepared from p-toluic acid and 1,2,4-trimethoxybenzene via the modification of known sequences of reactions^{13–17} involving intermediates **11–19** (schemes given in the Supporting Information).

The gem-dimethylcyclohexanone 9 was prepared (Scheme 2) by the conjugate addition of a methyl group¹⁸ to the allylcyclohexenone 21,19 obtained in turn from 20, the methyl ester analogue of Hagemann's ester. 20 Barbier reaction of 10a-c with 9 was initially carried out under sonication according to the method of Abad et al. 12a The resulting crude mixture furnished the expected desired products (8a-c) in low yield, along with a lot of undesired byproducts including aromatic dimeric compounds. Switching on to reaction without sonication improved the yield of the desired products, isolated in pure form as single diastereoisomers as determined by spectral analysis. The trans OH/H stereochemistry in the products, deduced by analogy, 10 received support from the X-ray structure determination of 24b subsequently done.21 Thus, the Barbier reaction of 10a-c with 9 appears to proceed diastereoselectively, in line with the addition of Grignard or alkyllithium reagents to 2-alkylcyclohexanones where a steric approach control of the addition has been suggested^{12b} to determine the course of reaction.

Treatment of each of the benzylcyclohexanols **8a-c** with NBS in acetonitrile²² led to the corresponding unstable bromomethylfurans **23a-c** via formation of the bromonium intermediate **22** followed by the intramolecular participation of the hydroxy group as depicted in Scheme 2. Further reaction of these intermediates **23a-c** with an excess of NBS gave the respective aromatic ring brominated products **24a-c**.

⁽¹¹⁾ For examples of the intramolecular 7-exo Heck reaction, see the following articles and reviews and the references therein: (a) Gracias, V.; Moore, J. D.; Djuric, S. T. Tetrahedron Lett. 2004, 45, 417–420. (b) Tietze, L. F.; Sommer, K. M.; Schneider, G.; Tapolcsányi, P.; Wölfling, J.; Mûller, P.; Noltemeyer, M.; Terlau, H. Synlett 2003, 1494–1496. (c) Santose, L. S.; Pilli, R. A. Synthesis 2002, 87–93. (d) Caddick, S.; Kofie, W. Tetrahedron Lett. 2002, 43, 9347–9350. (e) Link, J. T. Org. React. 2002, 60, 157–534. (f) Tietze, L. F.; Ila, H.; Bell, H. P. Chem. Rev. 2004, 104, 3453–3516. (g) Beletskaya, I. P.; Cheprakov, A. V. Chem. Rev. 2000, 100, 3009–3066. (h) Dounay, A. B.; Overman, L. E. Chem. Rev. 2003, 103, 2945–2963. (i) Link, J. T. In The Handbook of Organopalladium Chemistry for Organic Synthesis; Negishi, E., Ed.; Wiley, Sons: New York, 2002; Vol. 1, pp 1523–1560. (j) Link, J. T.; Overman, L. E. In Metal Catalyzed Cross Coupling Reactions; Diederich, F., Stang, P. J., Eds.; Wiley-VCH: Weinheim, 1998; pp 99–154.

^{(12) (}a) Abad, A.; Agulló, C.; Arnó, M.; Cuñat, A. C.; Meseguer, B.; Zaragozá, R. J. J. Org. Chem. **1998**, 63, 5100–5106. (b) Ashby, E. C.; Laemmle, J. T. Chem. Rev. **1975**, 75, 521–546.

⁽¹³⁾ Fleifel, A. M. J. Org. Chem. 1960, 25, 1024-1025.

⁽¹⁴⁾ Chakraborti, A. K.; Basak, A.; Grover, V. J. Org. Chem. 1999, 64, 8014–8017.

⁽¹⁵⁾ Sishido, K.; Goto, K.; Miyoshi, S.; Takaishi, Y.; Shibuya, M. J. Org. Chem. **1994**, *59*, 406–414.

⁽¹⁶⁾ Tori, M.; Hamaguchi, T.; Sagawa, K.; Sono, M.; Asakawa, Y. J. Org. Chem. **1996**, 61, 5362–5370.

⁽¹⁷⁾ Burnell, R. H.; Caron, S. Can. J. Chem. 1992, 1446–1454 [for an alternative method of preparation of 17].

⁽¹⁸⁾ Pal, S.; Mukhopadhyay, J. K.; Ghatak, U. R. J. Org. Chem.
1994, 59, 2687–2694.
(19) Cairns, P. M.; Howes, C.; Jenkins, P. R. J. Chem. Soc., Perkin

⁽¹⁹⁾ Cairns, P. M.; Howes, C.; Jenkins, P. R. J. Chem. Soc., Perkin Trans. 1 1990, 627–632.

⁽²⁰⁾ Begbie, A. L.; Golding, B. T. J. Chem. Soc., Perkin Trans. 1 1972, 602–605.

⁽²¹⁾ The ORTEP diagrams of **5a** and **24b** are given in the Supporting Information.

⁽²²⁾ Carreno, M. C.; Ruano, J. L. G.; Sanz, G.; Toledo, M. A.; Urbano, A. J. Org. Chem. **1995**, 60, 5328–5331.

SCHEME 2. Preparation of 24a-c

a: $R_1=R_2=R_3=H$; **b**: $R_1=R_3=H$, $R_2=Me$; **c**: $R_1=R_3=OMe$, $R_2=^{i}Pr$

SCHEME 3. Preparation of 7a-c

Regeneration of the allyl o-bromobenzylcyclohexanols (7a-c) from the bromomethylfuran derivatives (24a-c)was planned to be effected by using a zinc-mediated debromination fragmentation reaction.²³ Treatment of 24a and 24b with activated Zn dust and saturated NH₄-Cl solution indeed led to the allyl o-bromobenzyl cyclohexanols 7a and 7b in excellent yields (Scheme 3). In contrast, application of the same reaction to the highly substituted bromoaryl analogue 24c proved troublesome and resulted in the formation of the expected fragmentation product 7c along with the aromatic ring debrominated compound 8c. This was probably due to the buttressing effect²⁴ caused by the fully substituted phenyl ring in 24c, which makes the bromine atom relatively labile in comparison to that in the analogues **24a**-**b**. The aromatic ring debromination could be suppressed by reducing the reaction time to 6 h. Although the reaction was incomplete, remarkably the formation of 8c was not observed, and the desired 7c could be obtained in 65% yield (Scheme 3). However, fragmentation of 24c with acetic acid and activated zinc in methanol at room temperature²⁵ proved to be the method of choice, affording 7c in a substantially improved yield (84%). The

SCHEME 4. Preparation of 6a-c

assignment of *trans*-stereochemistry to the allyl *o*-bromobenzylcyclohexanols $\mathbf{7a} - \mathbf{c}$ and the corresponding precursors $\mathbf{8a} - \mathbf{c}$ was based on the established structure of $\mathbf{24b}$.

With the precursors 7a-c in hand, the synthetic exercise was next directed toward the Heck reaction. To our gratification, the intramolecular Heck cyclization of 7a and 7b with Pd(OAc)₂, K₂CO₃, and PPh₃ in acetonitrile under reflux²⁶ led to the seven-membered exo-cyclic alkenes **6a** and **6b** in excellent yields (Scheme 4). Not surprisingly, the highly crowded bromobenzyl cyclohexanol 7c produced under similar conditions only a small amount of the cyclized product 6c with most of the starting material remaining unchanged. However, the 7-exo product 6c could be obtained in 30% yield under forcing conditions (120 h reflux, 61% based on recovered starting material). Replacement of K₂CO₃ by triethylamine in acetonitrile under reflux for 120 h resulted in the formation of **6c** in 58% yield (95% based on recovered starting material). The yield of 6c could be further improved (Scheme 4) by using diisopropylethylamine as the base and heating the reaction mixture under reflux for 96 h (68%, 93% based on recovered starting material).

Finally, oxidative cleavage²⁷ of the *exo*-olefins **6a** and **6b** with OsO_4 and $NaIO_4$ gave the respective *trans*-hydroxy ketones **5a** and **5b** in good yields (Scheme 5). The structure of **5a** was confirmed by X-ray diffraction analysis.²²

⁽²³⁾ Tanino, K.; Onuki, K.; Asano, K.; Miyashita, M.; Nakamura, T.; Takahashi, Y.; Kuwajima, I. J. Am. Chem. Soc. **2003**, 125, 1498-1500.

^{(24) (}a) Schlosser, M. Eur. J. Org. Chem. 2001, 3975–3984. (b) Westheimer, F. H. In Steric Effects in Organic Chemistry; Newman, M. S., Ed.; Wiley: New York, 1956; pp 523–555.

^{(25) (}a) Clive, D. L. J.; Yu, M.; Sannigrahi, M. J. Org. Chem. **2004**, 69, 4116–4125. (b) Iguchi, M.; Nishiyama, A.; Etoh, H.; Okamoto, K.; Yamamura, S.; Kato, Y. Chem. Pharm. Bull. **1986**, 34, 4910–4915.

⁽²⁶⁾ Ma, S.; Ni, B. J. Org. Chem. 2002, 67, 8280-8288.

⁽²⁷⁾ Clive, D. L. J.; Wang, J. J. Org. Chem. 2004, 69, 2773–2784.

SCHEME 5. Preparation of 5a-c

SCHEME 6. Preparation of 3

Similar oxidative cleavage²⁷ of the highly crowded *exo*olefin **6c** produced the corresponding hydroxy ketone **5c**, albeit only in moderate yield (Scheme 5). Although it appeared that **5c** exists in equilibrium with the hemi-ketal form **25** in solution as revealed by the 1 H and 13 C NMR spectra, 28 direct oxidation 29 of this mixture of **5c** and **25** with Ag(II)O in dilute HNO₃ afforded (\pm)-komaroviquinone (**3**) as the only isolable product (Scheme 6).

With an expedient synthesis of (\pm)-komaroviquinone (3) accomplished, the remaining task of converting the key intermediate hydroxycycloheptenone $\bf 5b$ to (\pm)-faveline methyl ether ($\bf 1a$) was undertaken. While reaction of $\bf 5b$ with BF₃·Et₂O in CH₂Cl₂ at 0 °C^{5c} led (Scheme 7) only to the olefin $\bf 26$ (isofaveline methyl ether), the use of SOCl₂ in pyridine³⁰ (at -10 °C for 5 h) produced a mixture of the known tetrasubstituted olefinic ketone $\bf 27^{31}$ along with $\bf 26$ in a ratio of 2:1 (as shown by ¹H NMR). Finally, dehydration of $\bf 5b$ with KHSO₄ ^{8b} furnished an easily separable mixture of (\pm)-faveline methyl ether ($\bf 1a$) and its isomer $\bf 26$ (in 2:3 ratio as shown by ¹H NMR).

The present strategy for the synthesis of faveline methyl ether (1a) thus affords better overall yield of the product (4.48%) compared to that (1.95%) of Majetich et al.^{5c} and is also more flexible in approach to permit synthesis of many other related natural products.

(29) Snyder, C. D.; Rapoport, H. J. Am. Chem. Soc. **1972**, 94, 227–

(31) Ho, T. L.; Chen, C. K. Tetrahedron 1995, 51, 5819-5824.

SCHEME 7. Dehydration of 5b

Conclusion

The facile intramolecular Heck reaction of the olefinic intermediates $\mathbf{7a-c}$ provides a simple convergent route to rearranged diterpenoids incorporating a tricyclic dibenzo[a,d]cycloheptenone skeleton and led to the first total synthesis of (\pm)-komaroviquinone (3) through (\pm)-coulterone dimethyl ether ($\mathbf{5c}$). The methodology also constitutes a convenient alternative route to (\pm)-faveline methyl ether ($\mathbf{1a}$).

Experimental Section

2-Allyl-3,3-dimethylcyclohexanone (9). To a stirred suspension of CuI^{18} (14 g, 73.5 mmol) in dry Et_2O (86 mL) was added MeLi (75 mL, 1.6 M solution in Et₂O, 120 mmol) dropwise during 30 min (-25 °C). The resulting yellow suspension was cooled to -50 °C, and BF₃·OEt₂ (10.2 mL, 80.5 mmol) was added. After 10 min, the temperature was raised to -30 °C, a solution of **21** (3 g, 20 mmol) in Et₂O (25 mL) added dropwise during 1 h, and stirring was continued for 30 min at −30 °C. An additional aliquot of BF₃•OEt₂ (10.2 mL, 80.5 mmol) was added to the mixture and stirring continued for 2 h at -30 °C. The temperature of the bath was allowed to warm to -10 °C, and thereafter the reaction was quenched by dropwise addition of saturated NH₄Cl solution. The reaction mixture was filtered and the residue thoroughly washed with Et₂O. The combined organic extracts were washed with saturated NH₄Cl solution, saturated Na₂S₂O₃ solution, and brine and dried. After removal of the solvent, the crude product was purified by column chromatography over silica gel (100-200 mesh) using 2-4% EtOAc in hexane as eluent to afford **9** (3 g, 90%) as a colorless oil: IR (neat) ν 1710 cm⁻¹; ¹H NMR $(300 \text{ MHz}, \text{CDCl}_3) \delta 0.79 \text{ (s, 3H)}, 1.07 \text{ (s, 3H)}, 1.57 - 1.66 \text{ (m, })$ 2H), 1.71-1.92 (m, 2H), 2.03-2.11 (m, 1H), 2.21-2.36 (m, 3H), 2.40-2.47 (m, 1H), 4.90-4.98 (m, 1H), 5.02-5.03 (m, 1H), 5.70-5.83 (m, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 21.7, 22.8, 28.3, 29.2, 38.9, 39.4, 41.0, 60.6, 114.9, 137.6, 212.2; EI-MS m/z 166 (M⁺, 66), 111 (61), 95 (84), 69 (100). Anal. Calcd for C₁₁H₁₈O: C, 79.46; H, 10.91. Found: C, 79.32; H, 11.08.

(1S*,2S*)-2-Allyl-1-(3-methoxybenzyl)-3,3-dimethylcy-clohexanol (8a). To a well-stirred suspension of small chips of Li (590 mg, 85 mmol) in dry THF (160 mL) was added dropwise a mixture of $\bf 9$ (2.6 g, 15.7 mmol) and $\bf 10a$ (1.4 g, 8.9 mmol) in dry THF (50 mL) during 4 h at -10 °C. The reaction mixture was then stirred at 0 °C for 24 h. It was filtered and quenched with cold saturated solution of NH₄Cl. The layers were separated, and the organic part was evaporated under reduced pressure; the residue was taken in EtOAc (50 mL). The aqueous part was extracted with EtOAc (2 \times 20 mL); the

⁽²⁸⁾ The 1H NMR spectrum of the hydroxy ketone $\bf 5c$ in CDCl $_3$ solution showed additional peaks (see the Experimental Section) indicating the possible presence of two equilibrating forms. Recording the spectrum of a freshly prepared solution of $\bf 5c$ in C_5D_5N , however, suggested the presence of only one form ($\bf 5c$). The ^{13}C NMR spectrum was recorded later to establish if the keto ($\bf 5c$) or the hemiketal form ($\bf 25$) was present. Unfortunately, conversion to the other form ($\bf 25$) had taken place in the meantime and the solution showed peaks both for a carbonyl carbon ($\bf \delta$ 207.4) and a hemiketal carbon ($\bf \delta$ 102.9), with the former about four times more intense than the latter. The 1H NMR spectrum of the same solution, recorded again, now showed additional peaks supporting the presence of both the forms. The exchange appears to occur at a faster rate in CDCl $_3$ solution (where the presence of traces of acid is not unexpected) and after a week's time, the minor form appeared to have become the major one.

⁽³⁰⁾ Allen, W. S.; Bernstein, S. J. Am. Chem. Soc. **1955**, 77, 1028–1032.

combined organic extracts were washed with brine, dried, and concentrated. Column chromatography of the residue over silica gel (100–200 mesh) using 2–5% EtOAc in hexane yielded $\bf 8a$ (1.7 g, 70%) as a colorless oil: IR (neat) ν 3573, 2934, 1597, 1488, 1460, 1367, 1262, 1155 cm $^{-1}$; $^{1}{\rm H}$ NMR (300 MHz, CDCl $_{3}$) δ 0.94 (s, 3H), 1.01 (s, 3H), 1.15–1.24 (m, 2H), 1.28–1.46 (m, 3H), 1.51–1.65 (m, 2H), 2.21–2.29 (m, 1H), 2.38 (d, J=13 Hz, 1H), 2.50–2.58 (m, 1H), 3.12 (d, J=13 Hz, 1H), 3.79 (s, 3H), 4.96 (ddd, J=2, 3, 10 Hz, 1H), 5.09 (ddd, J=2, 3, 17 Hz, 1H), 5.89–6.02 (m, 1H), 6.72–6.80 (m, 3H), 7.21 (t, J=8 Hz, 1H); $^{13}{\rm C}$ NMR (75 MHz, CDCl $_{3}$) δ 17.7, 21.5, 30.1, 32.2, 35.0, 38.0, 41.9, 48.1, 53.0, 54.9, 74.4, 111.5, 113.8, 116.3, 123.0, 128.9, 138.7, 142.0, 159.3; EI-MS m/z 288 (M $^{+}$, 2), 273 (11), 167 (100). Anal. Calcd for $\rm C_{19}H_{28}O_{2}$: C, 79.12; H, 9.78. Found: C, 79.31; H, 9.65.

 $(2S^*,3aS^*,7aS^*)$ -7a-(2-Bromo-5-methoxybenzyl)-2-bromomethyl-4,4-dimethyloctahydrobenzofuran (24a). To a well-stirred solution of 8a (1.17 g, 4.06 mmol) in dry CH₃CN $(20\ mL)$ was added NBS^{22} $(0.795\ g,\,4.46\ mmol)$ portionwise at −20 °C. The reaction mixture was stirred for 2 h at the same temperature. Then another lot of NBS (0.795 g, 4.46 mmol) was added and the reaction mixture further stirred for 2 h at -20 °C. Aqueous Na₂S₂O₃ solution (10%) was then added, and the layers were separated. The organic part was evaporated under reduced pressure, and the crude product was extracted with Et₂O. The Et₂O extract was washed with Na₂S₂O₃ solution (5%, 50 mL) and brine, dried, and then evaporated to leave an oil. The crude oil was chromatographed over silica gel (100-200 mesh) using 1-2% EtOAc in hexane to yield 24a (1.6 g, 88%) as an oil, which solidified after cooling: mp 70-71 °C; IR (KBr) v 2958, 1590, 1467 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 0.95 (s, 3H), 1.17 (s, 3H), 1.14–1.25 (m, 3H), 1.34– 1.45 (m, 1H), 1.47-1.62 (m, 1H), 1.67-1.82 (m, 1H), 1.85-1.94 (m, 2H), 2.08-2.21 (m, 1H), 2.81 (d, J = 14 Hz, 1H), 3.30(dd, J = 7, 10 Hz, 1H), 3.40 (dd, J = 5, 10 Hz, 1H), 3.52 (d, J = 7, 10 Hz, 1H), 3.5J = 14 Hz, 1H, 3.78 (s, 3H), 4.15-4.22 (m, 1H), 6.64 (dd, J = 1.00 (m, 1H), 6.64 (dd, J = 13, 9 Hz, 1H), 7.16 (d, J = 3 Hz, 1H), 7.39 (d, J = 9 Hz, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 19.9, 28.8, 30.5, 31.1, 32.6, 33.2, 33.7, 37.6, 44.5, 53.1, 55.3, 74.3, 85.0, 114.2, 116.6, 118.6, 132.8, 138.9, 158.1; EI-MS m/z 448 (M⁺ for 2 × Br⁸¹, 0.02), 446 (M⁺ for Br^{79} and Br^{81} , 0.04), 444 (M⁺ for $2 \times Br^{79}$, 0.02), 247 (66), 245 (62), 69 (100). Anal. Calcd for $C_{19}H_{26}Br_2O_2$: C, 51.14; H, 5.87. Found: C, 51.06; H, 5.99.

 $(2S^*,3aS^*,7aS^*)$ -7a-(2-Bromo-4-isopropyl-3,5,6-trimethoxybenzyl)-2-bromomethyl-4,4-dimethyloctahydroben**zofuran (24c).** To a well-stirred solution of **8c** (2 g, 5.13 mmol) in dry CH₃CN (170 mL) was added NBS²² (1.1 g, 6.18 mmol) portionwise at -20 °C. The reaction mixture was stirred for 3 h at the same temperature. Then an additional lot of Nbromosuccinimide (1.1 g, 6.18 mmol) was added and stirring continued for another 48 h at room temperature. After usual workup, the product was chromatographed over silica gel (100-200 mesh) using 1-2% EtOAc in hexane to yield 24c (2.39 g, 85%) as a white solid: mp 116-118 °C; IR (KBr) ν 2955, 1581, 1450 cm $^{-1}$; ¹H NMR (300 MHz, CDCl₃) δ 0.83 $^{-1}$ $0.92\,(m,\,1H),\,0.97\,(s,\,3H),\,1.18\,(s,\,3H),\,1.26-1.46\,(m,\,1H),\,1.31$ (d, J = 7 Hz, 6H), 1.51 - 1.74 (m, 3H), 1.75 - 1.83 (m, 1H), 1.93 -2.19 (m, 3H), 3.01 (d, J = 13 Hz, 1H), 3.21 (t, J = 9 Hz, 1H),3.35-3.49 (m, 3H), 3.73 (s, 3H), 3.76 (s, 3H), 3.84 (s, 3H), 4.08-4.15 (m, 1H); 13 C NMR (75 MHz, CDCl₃) δ 19.6, 20.9, 22.0, 26.8, 28.6, 31.2, 31.8, 32.5, 32.9, 33.6, 36.6, 39.6, 53.2, 59.4, 60.0, 61.3, 75.2, 85.5, 116.8, 131.1, 134.5, 149.3, 151.0, 151.7; EI-MS m/z 550 (M⁺ for 2 × Br⁸¹, 1), 548 (M⁺ for Br⁷⁹ and Br⁸¹, 3), 546 (M⁺ for $2 \times Br^{79}$, 1), 533 (3), 505 (4), 245 (100). Anal. Calcd for C₂₄H₃₆Br₂O₄: C, 52.57; H, 6.62. Found: C, 52.66; H,

(1 S^* ,2 S^*)-2-Allyl-1-(2-bromo-5-methoxybenzyl)-3,3-dimethylcyclohexanol (7a). A mixture of 24a (1 g, 2.24 mmol), zinc²³ (2.5 g, 38.2 mmol), and NH₄Cl solution (10 mL, 50%) in Et₂O (15 mL) was vigorously stirred at room temperature for 6 h. Then another lot of zinc (2.5 g, 38.2 mmol) was added, and the reaction mixture was stirred for 6 h. The mixture was

diluted with Et₂O (100 mL) and filtered. The organic part was washed with brine, dried, and concentrated. The crude product was chromatographed over silica gel (100–200 mesh) using 2–5% EtOAc in hexane to provide **7a** (720 mg, 88%) as white solid: mp 62–63 °C; IR (KBr) ν 3523, 2936, 1587 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 0.95 (s, 3H), 1.01 (s, 3H), 1.23–1.62 (m, 9H), 2.24–2.31 (m, 1H), 2.48–2.58 (m, 1H), 2.65 (d, J = 13 Hz, 1H), 3.27 (d, J = 13 Hz, 1H), 3.77 (s, 3H), 4.98 (dd, J = 1, 10 Hz, 1H), 5.12 (dd, J = 1, 17 Hz, 1H), 5.90–6.04 (m, 1H), 6.65 (dd, J = 3, 9 Hz, 1H), 6.84 (d, J = 3 Hz, 1H), 7.42 (d, J = 9 Hz, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 17.7, 21.7, 30.2, 32.2, 35.2, 37.4, 41.9, 46.9, 53.7, 55.4, 76.1, 113.8, 114.1, 116.8, 117.9, 133.4, 138.4, 141.8, 158.5; EI-MS m/z 368 (M⁺ for Br⁸¹, 19), 366 (M⁺ for Br⁷⁹, 11), 353 (91), 351 (100). Anal. Calcd for C₁₉H₂₇BrO₂: C, 62.13; H, 7.41. Found: C, 62.29; H, 7.56.

 $(1S^*,2S^*)$ -2-Allyl-1-(2-bromo-4-isopropyl-3,5,6-trimethoxybenzyl)-3,3-dimethylcyclohexanol (7c). Method A. A mixture of **24c** (1 g, 1.82 mmol), activated zinc²³ (4.47 g, 68.4 mmol), NH₄Cl solution (40 mL, 50%), and ether (40 mL) was vigorously stirred at room temperature for 6 h. The reaction mixture was diluted with ether (100 mL) and filtered. After usual workup, the crude residue was flash chromatographed over silica gel (230-400 mesh); 0.5-1% EtOAc in hexane eluted 7c (554 mg, 65%) as a colorless oil (which solidified on cooling) while 2-4% EtOAc in hexane eluted the starting material 24c (275 mg). 7c: mp 80-81 °C (crystallized from MeOH); IR (KBr) ν 3484, 2935, 1452 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 0.83-1.00 (m, 1H), 0.94 (s, 3H), 1.00 (s, 3H), 1.12-1.68 (m, 7H), 1.31 (d, J = 7 Hz, 3H), 1.33 (d, J = 7 Hz, 3H),2.28-2.32 (m, 1H), 2.44-2.54 (m, 1H), 2.85 (d, J = 14 Hz, 1H), 3.32 (d, J = 14 Hz, 1H), 3.38 - 3.50 (m, 1H), 3.74 (s, 3H), 3.81(s, 3H), 3.86 (s, 3H), 4.96 (d, J = 10 Hz, 1H), 5.10 (d, J = 17Hz, 1H), 5.96–6.05 (m, 1H); $^{13}\mathrm{C}$ NMR (75 MHz, CDCl $_3$) δ 18.1, 21.6, 21.9, 22.0, 26.8, 30.6, 32.2, 35.4, 37.5, 42.1, 43.0, 54.4,59.8, 60.5, 61.4, 76.7, 113.6, 116.5, 130.8, 134.9, 142.5, 148.3,151.7, 151.9; EI-MS m/z 470 (M⁺ for Br⁸¹, 2), 468 (M⁺ for Br⁷⁹, 2), 304 (100). Anal. Calcd for C₂₄H₃₇BrO₄: C, 61.40; H, 7.94. Found: C, 61.29; H, 8.06.

Method B. To a solution of **24c** (20 mg, 0.04 mmol) in MeOH (0.4 mL) were added freshly activated zinc (20 mg, 0.31 mmol) and glacial acetic acid (0.3 mL), and the mixture was stirred at room temperature. ²⁵ After 6 h, an additional lot of freshly activated zinc (30 mg, 0.47 mmol) and MeOH (0.4 mL) was added, and the stirring was continued for another 24 h. The reaction mixture was filtered, diluted with H₂O, and extracted with EtOAc (2 \times 10 mL). The combined organic extracts were washed with saturated NaHCO₃ solution and brine, dried, and concentrated. The crude product was purified by column chromatography over silica gel (100–200 mesh) using 2–4% EtOAc in hexane to furnish **7c** (14.5 mg, 84%).

 $(4aS^*,11aS^*)$ -7-Methoxy-1,1-dimethyl-10-methylene-1,2,3,4,5,10,11,11a-octahydrodibenzo[a,d]cyclohepten-4aol (6a). A mixture of 7a (500 mg, 1.36 mmol), K₂CO₃ (1.13 g, 8.2 mmol), and PPh₃ (140 mg, 0.53 mmol) in dry CH₃CN (50 mL) was degassed by sonication for 30 min. Pd(OAc)2 (60 mg, 0.267 mmol) was added to the reaction mixture. The mixture was stirred for 1 h at room temperature and then heated to reflux²⁶ for 24 h under nitrogen atmosphere. The mixture was cooled and filtered through Celite, and the residue washed with dry ethyl acetate. The combined filtrate was evaporated under reduced pressure to leave a residue. The crude product was chromatographed over silica gel (100-200 mesh) using 2% EtOAc in hexane to afford 6a (325 mg, 83%) as a colorless oil: IR (neat) ν 3473, 2932, 1606 cm⁻¹; ¹H NMR (300 MHz, CD_3OD) δ 0.94 (s, 3H,), 0.96 (s, 3H,), 1.23–1.53 (m, 6H), 1.67– 1.73 (m, 1H), 1.90-1.94 (m, 1H), 2.36-2.45 (m, 1H), 2.53 (d, J = 13 Hz, 1H), 2.64 (d, J = 14 Hz, 1H), 2.76 (d, J = 14 Hz, 1H), 3.77 (s, 3H), 4.96 (d, J = 2 Hz, 1H), 5.07 (s, 1H), 6.65 (d, J 3 Hz, 1H), 6.73 (dd, J = 3, 8 Hz, 1H), 7.10 (d, J 8 Hz, 1H); 13 C NMR (75 MHz, CD₃OD) δ 19.6, 22.0, 32.4, 32.8, 35.1, 42.1, 43.3, 52.7, 55.6, 56.9, 72.8, 112.3, 112.6, 117.6, 129.3, 137.9,

138.4, 153.9, 160.3; EI-MS $\it m/z$ 286 (M $^+$, 53), 202 (64), 167 (100). Anal. Calcd for $\rm C_{19}H_{26}O_2$: C, 79.68; H, 9.15. Found: C, 79.75; H, 9.21.

 $(4aS^*,11aS^*)$ -8-Isopropyl-6,7,9-trimethoxy-1,1-dimethyl-10-methylene-1,2,3,4,5,10,11,11a-octahydrodibenzo-[a,d]cyclohepten-4a-ol (6c). Method A. A mixture of 7c (40 mg, 0.09 mmol), K₂CO₃ (70.6 mg, 0.51 mmol), and PPh₃ (8.9 mg, 0.03 mmol) in dry CH₃CN (5 mL) was degassed by sonication for 30 min. Pd(OAc)2 (3.8 mg, 0.02 mmol) was added to the reaction mixture, and the mixture stirred for 1 h at room temperature. It was then heated²⁶ at 90 °C for 120 h. Usual workup gave a crude product which was chromatographed over silica gel (60–120 mesh). Eluting with 1–2% EtOAc in hexane afforded starting material 7c (20 mg), while 2-4% EtOAc in hexane furnished the cyclized product 6c (10 mg, 30%; 61%) based on recovered starting material) as a thick oil, which solidified on cooling: mp 124-126 °C (crystallized from dry methanol); IR (KBr) ν 3478, 2936, 1581 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 0.92 (s, 3H), 1.01 (s, 3H), 1.22–1.35 (m, overlapped, 2H), 1.32 (d, J=7 Hz, 3H), 1.34 (d, J=7 Hz, 3H,), 1.43-1.56 (m, 3H), 1.64 (s, 1H), 1.78-2.04 (m, 3H), 2.46 (d, J = 14 Hz, 1H), 2.69 (brd, J = 13 Hz, 1H), 3.14 (d, J = 14)Hz, 1H), 3.45 (sep, J = 7 Hz, 1H), 3.61 (s, 3H), 3.76 (s, 3H), 3.84 (s, 3H), 5.08 (s, 1H), 5.35 (s, 1H); ¹³C NMR (75 MHz, $CDCl_3$) δ 18.7, 21.4, 22.07, 22.10, 25.9, 32.1, 33.1, 34.3, 41.6, 42.0, 42.2, 59.7, 60.4, 60.5, 61.0, 70.7, 115.5, 126.8, 133.9, 145.8, 148.5, 150.5, 151.0 [one carbon signal could not be identified due to overlap]; EI-MS m/z 388 (M+, 100), 247 (11), 224 (16). Anal. Calcd for C₂₄H₃₆O₄: C, 74.19; H, 9.34. Found: C, 74.06; H, 9.27.

Method B. A mixture of **7c** (280 mg, 0.60 mmol), triethylamine (362.5 mg, 0.5 mL, 3.58 mmol), and PPh $_3$ (62.6 mg, 0.24 mmol) in dry acetonitrile (25 mL) was sonicated for 30 min. Palladium acetate (26.8 mg, 0.12 mmol) was added to the reaction mixture and the mixture stirred for 1 h at room temperature. It was then heated at 90 °C for 120 h. After usual workup, the crude product was chromatographed over silica gel (60–120 mesh) to afford the starting material **7c** (110 mg) and **6c** (134 mg, 58%; 95% based on recovered starting material).

Method C. A mixture of **7c** (445 mg, 0.95 mmol), diisopropylethylamine (1.23 g, 1.65 mL, 9.49 mmol), and PPh₃ (99.5 mg, 0.38 mmol) in dry CH₃CN (40 mL) was sonicated for 30 min. Pd(OAc)₂ (42.6 mg, 0.19 mmol) was added, and the mixture was stirred for 1 h at room temperature. It was then heated at 90 °C for 96 h. After usual workup, the crude product was chromatographed over silica gel (60–120 mesh) to afford **7c** (120 mg) and **6c** (250 mg, 68%; 93% based on recovered starting material).

 $(4aS^*,11aS^*)-4a$ -Hydroxy-7-methoxy-1,1-dimethyl-1,2,-3,4,4a,5,11,11a-octahydrodibenzo[a,d]cyclohepten-10one (5a). To a solution of 6a (200 mg, 0.7 mmol) in a mixture of BuOH (9 mL), acetone (9 mL), and H2O (9 mL) was added N-methylmorpholine oxide (495 μ L, 50% aqueous solution). Three small crystals of OsO₄ ²⁷ were then added at 0 °C. The mixture was stirred for 5 h at 0 °C and for 2 h at room temperature. A solution of Na₂SO₃ (14 mL, 45% aqueous solution) was added, and the reaction mixture was stirred for 45 min. The solid was filtered and the filtrate evaporated to leave a residue, which was taken up in EtOAc. The solid mass was dissolved in H₂O and extracted with EtOAc (50 mL). The combined organic extracts were washed with brine and dried, and solvent was removed in a rotary evaporator under reduced pressure to leave the crude trihydroxy compound as a white solid, which was directly used in the next step without purification.

To the solution of the white solid in MeOH (10 mL) was added a solution of NaIO₄ (prepared by dissolving 222 mg in 3 mL of water) dropwise at 0 °C. The reaction mixture was stirred at 0 °C for 2 h, then at room temperature for 3 h, and filtered. The residue was dissolved in $\rm H_2O$ and extracted with EtOAc (100 mL). The combined organic extracts were washed

with brine and then dried. The solvent was evaporated under reduced pressure to give a white solid. The crude solid was chromatographed over silica gel (60–120 mesh) using 10–15% ethyl acetate in hexane to afford $\bf 5a$ as a white solid (180 mg, 86%): mp 125–127 °C (crystallized from 5% EtOAc in hexane); IR (KBr) ν 3470, 2937, 1657 cm $^{-1}$; 1 H NMR (300 MHz, CDCl $_3$) δ 0.82 (s, 3H), 0.99 (s, 3H), 1.09–1.18 (m, 1H), 1.39–1.73 (m, 5H), 1.82–1.95 (m, 2H), 2.62 (d, J=14 Hz, 1H), 2.67 (d, J=9 Hz, 1H), 2.85 (2 × d, J=11, 18 Hz, 1H), 3.11 (d, J=14 Hz, 1H), 3.85 (s, 3H), 6.62 (d, J=2 Hz, 1H), 6.84 (brd, J=7 Hz, 1H), 7.70 (d, J=9 Hz,1H); 13 C NMR (75 MHz, CDCl $_3$) δ 18.5, 21.5, 32.1, 33.7, 38.5, 39.1, 40.7, 48.3, 49.9, 55.4, 73.4, 111.8, 115.9, 131.0, 131.3, 139.5, 162.5, 205.3; EI-MS m/z 288 (M $^+$, 28), 149 (100). Anal. Calcd for $\rm C_{18}H_{24}O_3$: C, 74.97; H, 8.39. Found: C, 75.09; H, 8.31.

(4aS*,11aS*)-4a-Hydroxy-8-isopropyl-6,7,9-trimethoxy-1,1-dimethyl-1,2,3,4,4a,5,11,11a-octahydrodibenzo[a,d]**cyclohepten-10-one (5c).** To a solution of **6c** (166 mg, 0.43 mmol) in a mixture of tBuOH (7 mL), acetone (7 mL), and H_2O (7 mL) was added N-methylmorpholine oxide (1 mL, 50% aqueous solution). Os O_4 ²⁷ ($\widetilde{22}$ mg, 0.866 mmol, 20 mol %) was then added at 0 °C. The mixture was stirred at 0 °C for 3 h and then at room temperature for 96 h. A solution of Na₂SO₃ (40 mL, 45% aqueous solution) was added to the reaction mixture, which was stirred for 1 h. After usual workup as described for 5a, the crude product obtained was chromatographed over silica gel (60-120 mesh). Elution with 2% EtOAc in hexane furnished the starting olefin 6c (76 mg), while 30% ethyl acetate in petroleum ether afforded the trihydroxy compound (96 mg) (as revealed from ¹H NMR and IR spectra), which was directly used in the next step.

To the solution of above trihydroxy compound (95 mg) in MeOH (7 mL) was added a solution of NaIO₄ (prepared by dissolving 175 mg in 2 mL of water) dropwise at 0 °C. The reaction mixture was stirred at 0 °C for 2 h and then at room temperature for 24 h and filtered. Usual workup furnished the crude product which was chromatographed over silica gel (60-120 mesh) using 10-15% EtOAc in hexane to afford 5c (70 mg, 42% in two steps; 77% based on recovered starting material 6c) as a white solid: mp 138-140 °C (crystallized from 2% ethyl acetate in petroleum ether); IR (KBr) ν 3484, 2944, 1673, 1574 cm $^{-1}$; 1 H NMR (300 MHz, pyridine- d_{5}) δ 0.85 (s, 3H), 1.07 (brd t, J= 12 Hz, 1H), 1.21 (s, 3H), 1.37 (d, J=7 Hz, 3H), 1.39 (d, J = 7 Hz, 3H), 1.35-1.48 (m, overlap, 2H), 1.61-1.71 (m, 1H), 1.80 (d, J = 11 Hz, 1H), 2.02 (d, J = 13Hz, 1H), 2.14-2.23 (m, 1H), 2.86 (d, J = 17 Hz, 1H), 3.06-103.10 (m, 1H), 3.21 (m, 1H), 3.39 (dd, J 12, 16 Hz, 1H), 3.58(sept, J 7 Hz, 1H), 3.77 (s, 3H), 3.88 (s, 6H), 5.80 (s, D₂O exchangeable, 1H); 13 C NMR (75 MHz, pyridine- d_5) [a mixture of **5c** (major) and the hemiketal **25** (minor); only the major peaks shown] δ 19.2, 22.02, 22.06, 22.09, 32.1, 34.1, 40.4, 40.6, 41.6, 60.4, 63.4, 72.5, 128.1, 134.3, 147.6, 151.8, 154.4, 207.4 [additional peaks at δ 78.6 and 102.8 are assignable to **25**]; EI-MS m/z 390 (M+, 73), 267 (29), 251 (100). Anal. Calcd for C₂₃H₃₄O₅: C, 70.74; H, 8.78. Found: C, 70.87; H, 8.83.

(\pm)-Komaroviquinone (3). To a stirred solution containing 5c and 25 (28 mg, 0.07 mmol) in dioxane (1.4 mL, redistilled over sodium) and silver(II) oxide²⁹ (36 mg, 0.29 mmol) was added HNO $_3$ (717 μ L, 6 N) at 5 °C. After the mixture was stirred for 4 h at 5 °C, another portion of silver(II) oxide (36 mg) and HNO₃ (717 µL, 6 N) was added and the stirring continued for 4 h at the same temperature. The mixture was diluted with water and extracted with EtOAc (50 mL). The organic extract was washed with brine, dried, and evaporated to give an orange oil. The product was purified over silica gel (60-120 mesh) using 5% EtOAc in hexane to give 3 (18 mg, 69%) as a yellow-orange solid: mp 123-124 °C [natural (+)-3 was reported⁴ as an orange oil]; \hat{IR} (KBr) ν 3404, 2944, 1651, 1606, 1455 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 0.86 (s, 3H), 0.95 (s, 3H), 1.21 (d, J = 7 Hz, 3H), 1.21 (d, J = 7 Hz, 3H), 1.11-1.25 (m, overlapped, 1H), 1.56-1.66 (m, 2H), 1.68-1.75 (m, 2H), 1.78-1.90 (m, 1H), 1.99-2.08 (m, 2H), 2.25 (d, J = 1.90)



19 Hz, 1H), 2.30 (dd, J=13, 9 Hz, 1H), 2.55 (d, J=19 Hz, 1H), 3.23 (septet, J=7 Hz, 1H), 3.98 (s, 3H), 5.98 (brs, 1H); 13 C NMR (75 MHz, CDCl₃) δ 15.6, 20.38, 20.42, 24.3, 27.0, 29.8, 30.3, 31.3, 32.0, 39.0, 45.7, 51.5, 61.1, 79.3, 100.9, 137.1, 138.9, 142.2, 156.1, 183.6, 189.1 [1 H NMR and 13 C NMR of (\pm)-3 are identical with those of the natural (+)-3 4]; EI-MS m/z 360 (M⁺, 16), 316 (100), 301 (40). Anal. Calcd for C₂₁H₂₈O₅: C, 69.98; H, 7.83. Found: C, 69.83; H, 7.96.

Reaction of 5b with $BF_3 \cdot OEt_2$: Formation of (11aS*)-7-Methoxy-1,1,8-trimethyl-1,2,3,5,11,11a-hexahydrodibenzo[a,d]cyclohepten-10-one (26). To a stirred solution of 5b (20 mg, 0.066 mmol) in dry CH₂Cl₂ (2 mL) was added BF₃· OEt_2 (10 μ L) at 0 °C. 5c The reaction mixture was stirred at 0 °C for 3 h and then at room temperature for 6 h. The reaction mixture was quenched with saturated NaHCO₃ solution (5 mL) and extracted with CH2Cl2 (20 mL). The organic part was washed with brine and dried and the solvent evaporated to give a yellow oil. The crude oil was purified by preparative TLC (multiple run in 2% ethyl acetate-hexane) to give 26 (9.3 mg, 49%) as the only isolable product as a white solid: mp 135-136 °C (crystallized from EtOAc in hexane); IR (KBr) ν 2916, 1654, 1601 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 0.80 (s, 3H), 0.99 (s, 3H), 1.37-1.39 (m, 2H), 2.03-2.19 (m, 3H), 2.19 (s, 3H), 2.59 (dd, J = 11, 17 Hz, 1H), 2.88 (d, J = 17 Hz, 1H), 3.31 (d, J = 15 Hz, 1H), 3.83 (d, J = 15 Hz, 1H), overlapped by3H singlet at 3.89), 3.89 (s, 3H), 5.67 (s, 1H), 6.60 (s, 1H), 7.73 (s, 1H); 13 C NMR (75 MHz, CDCl₃) δ 15.6, 19.2, 23.1, 29.4, 32.4, 37.2, 42.8, 43.1, 43.7, 55.5, 110.3, 121.9, 125.2, 128.8, 131.8, 135.6, 140.4, 161.6, 203.0; EI-MS m/z 284 (M⁺, 61), 256 (97), 200 (100). Anal. Calcd for C₁₉H₂₄O₂: C, 80.24; H, 8.51. Found: C, 80.39; H, 8.46.

Reaction of 5b with SOCl₂ in Pyridine: Formation of 26 and 7-Methoxy-1,1,8-trimethyl-1,2,3,4,5,11-hexahydrodibenzo[a,d]cyclohepten-10-one (27). To a stirred solution of 5b (20 mg, 0.07 mmol) in pyridine (714 μ L) was added SOCl₂ ³⁰ (72 μ L, 0.99 mmol) at -10 °C. The reaction mixture was stirred for 5 h at the same temperature. The reaction mixture was diluted with Et₂O (25 mL), and the organic part was washed successively with HCl (10 mL, 5%), water, and brine, and dried. The solvent was evaporated to give a mixture

of **26** and **27** as an oil. The crude oil was purified by preparative TLC (multiple run in 2% EtOAc—hexane) to give **26** (5.6 mg, 30%) and **27**³¹ (6 mg, 32%) as an oil.

Reaction of 5b with KHSO₄: Formation of (\pm)-Faveline Methyl Ether (1a) and 26. The keto alcohol 5b (40 mg, 0.132 mmol) was fused with KHSO₄ 8 (40 mg, 0.294 mmol) at 190 $^\circ$ C under N₂ atmosphere for 45 min, and the resulting mixture was sublimed at 140–145 $^\circ$ C (0.1 mm of Hg) to give a solid. The product was purified by preparative TLC (multiple run in 2% EtOAc—hexane) to give 1a (10.8 mg, 28%) and 26 (15.8 mg, 42%) as white solids.

1a: mp 137–138 °C [lit.8b mp 138–140 °C]; IR (KBr) ν 2929, 1655, 1600 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 0.75 (s, 3H), 1.12 (s, 3H), 1.42–1.62 (m, 2H), 1.64–1.76 (m, 2H), 2.18 (s, 3H), 2.30–2.38 (m, 3H), 3.01–3.03 (m, 2H), 3.87 (s, 3H), 6.29 (s, 1H), 6.59 (s, 1H) 7.62 (s, 1H); ¹³C NMR (75 MHz, CDCl₃) δ 15.5, 20.7, 25.1, 28.9, 38.4, 40.2, 42.5, 43.0, 51.3, 55.4, 112.9, 124.7, 125.5, 129.7, 131.2, 136.3, 147.7, 160.9, 201.3; EI-MS m/z 284 (M⁺, 100), 215 (81), 202 (94). Anal. Calcd for C₁₉H₂₄O₂: C, 80.24; H, 8.51. Found: C, 80.11; H, 8.65.

Acknowledgment. We are grateful to Prof. U. R. Ghatak of this institute for his suggestions and criticism and Dr. N. Uchiyama of Kyoto Pharmaceutical University for providing copies of spectra of natural (+)-komaroviquinone for comparison. We also thank UGC (New Delhi) and Dabur Research Foundation (India) for the award of a Research Fellowship to S.S. and the EPSRC and the University of Reading for funds of the Image Plate system.

Supporting Information Available: Schemes 2 and 3 (containing compounds 11–19); experimental procedures for 5b, 6b, 7b, 8b,c, 10b,c, 12–19, 20a, 21, and 24b; ¹H NMR and ¹³C NMR spectral data of compounds 1a, 3, 5a–c, mixture of 5c and 25, 6a–c, 7a–c, 8a–c, 24a–c, and 26; IR spectral data of 3; and ORTEP diagrams of 5a and 24b. This material is available free of charge via the Internet at http://pubs.acs.org.

JO051082I